# Reactor, Boiler & Auxiliaries - Course 233 HEAT TRANSPORT H2 ADDITION

#### I. PURPOSE OF SYSTEM

The purpose of this system is to minimize the amount of dissolved  ${\rm O}_2$  gas in the HT system. The  ${\rm O}_2$  is a product of radiolysis of the HT  ${\rm D}_2{\rm O}$  caused by fast neutrons and rays inside the reactor.

The reversible radiolysis reaction (simplified) is:

The  $\rm D_2$  and  $\rm O_2$  gas produced will be in solution at the normal operating pressure of the HT system. H<sub>2</sub> gas is added to the HT D<sub>2</sub>O to drive this reversible reaction to the left, and hence to reduce the O<sub>2</sub> concentration (scavenge the O<sub>2</sub>) in the coolant to a tolerable limit (a few  $\mu g$  O<sub>2</sub>/kg D<sub>2</sub>O). An excess of dissolved H<sub>2</sub> must be avoided to prevent embrittlement of the Zircaloy pressure tubes.

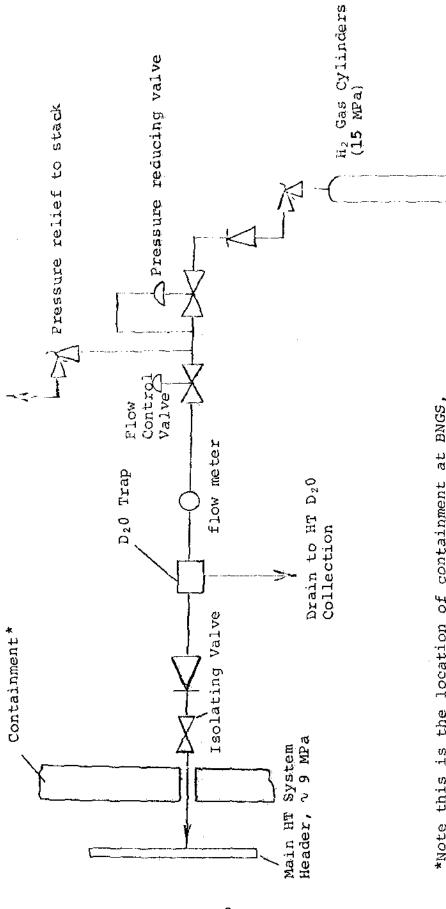
Oxygen control by  $\rm H_2$  addition is one of the most important chemical controls of the HT system. Dissolved oxygen promotes corrosion and hence crud formation and activated crud deposits cause undesirable high fields in the HT system. A longer term problem is wastage of system components due to corrosion.

#### II. SYSTEM DESCRIPTION

#### 1. Choice of Gas

Because  $\rm D_2$  and  $\rm H_2$  behave identically as far as reaction (1) is concerned, either deuterium or hydrogen gas additions would scavenge the dissolved oxygen equally well.

Two economic considerations contribute to the choice between  ${\rm H}_2$  and  ${\rm D}_2$ :



\*Note this is the location of containment at BNGS, at PNGS containment is between the trap and flowmeter.

Figure 1 HT System H2 Gas Addition

- (a) Favouring H2: D2 cost much more than H2.
- (b) Favouring D<sub>2</sub>: Use of H<sub>2</sub> downgrades HT isotopic (by typically  $\sim 0.3\%$  year). Recall that minimum acceptable HT isotopic is  $\sim 97\%$ .

## 2. Equipment

A typical  $\rm H_2$  addition system is shown in Figure 1, and consists of high pressure (15 MPa)  $\rm H_2$  cylinders, a pressure reducing valve, a flow control valve and check valve.

The pressure reducing valve maintains the  $\rm H_2$  addition pressure at HT system pressure,  $^{\circ}9$  MPa. Flow is controlled by a flow control valve (usually manual), and measured by a flow meter. A check valve prevents backflow of HT D2O into the gas system. D2O leakage past the check valve is collected in a D2O trap, which can be drained to the HT D2O collection system. Addition points to the HT system are usually at the main headers. Pressure relief should be provided on the gas bottles and also downstream of the pressure reducing valve.

Notice that this system is a place where the HT system penetrates containment. In the case of a piping rupture and subsequent loss of coolant this penetration would not be considered very large (the gas addition lines are typically  $\sim 1$  cm diameter).

### III. OPERATING FEATURES

#### (a) HT H<sub>2</sub> Concentration Control

The addition rate of  $\rm H_2$  in this system is normally adjusted manually to maintain the  $\rm H_2$  concentration (and hence the  $\rm O_2$  concentration) within the specification, typically between 5 - 25 cc  $\rm H_2/kg$   $\rm D_2O_*(1)$ 

As  $\mathrm{O}_2$  is difficult to measure, and hence control directly, it is the  $\mathrm{H}_2$  concentration that is controlled to provide an indirect control of  $\mathrm{O}_2$  concentrations. Periodic  $\mathrm{O}_2$  measurement will confirm that the control by  $\mathrm{H}_2$  addition is adequate.

<sup>(</sup>i) The desired  ${\rm H_2}$  concentration is usually  ${\sim}\,10\,{\rm cc}$   ${\rm H_2/kg}$   ${\rm D_2O}$ .

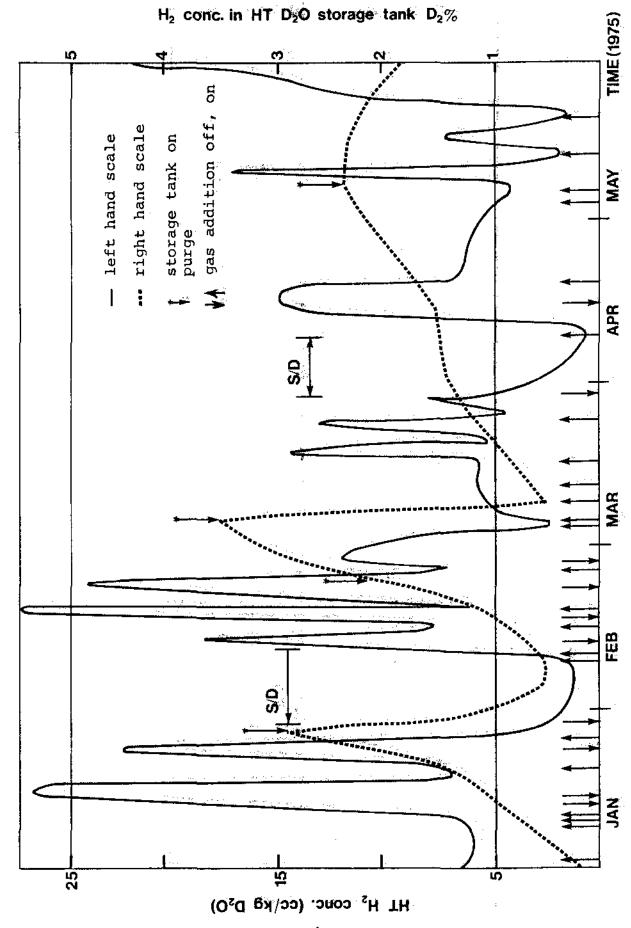


Figure 2 - Average Daily H<sub>2</sub> concentration measurements PNGS-A Unit 1

-- 4 --

A typical graph showing HT H<sub>2</sub> concentration is given in Figure 2 for PNGS-A Unit 1 during 1975. Manual adjustment of the H<sub>2</sub> gas addition flow control valve is shown as  $\uparrow$  for a valve opening increase and  $\downarrow$  for a valve opening decrease. Average H<sub>2</sub> concentration for the period is  $\sim \!\! 10$  cc dissolved H<sub>2</sub>/kg D<sub>2</sub>O. The control range between 5 - 25 cc H<sub>2</sub>/kg D<sub>2</sub>O is indicated on Figure 2. The effect of the H<sub>2</sub> flow rate changes on the concentration is clearly seen.

# (b) Degassing of HT H2 Into HT System Components

Not all the HT system equipment is under high pressure and there is HT equipment at lower pressures where dissolved  $\rm H_2$  is able to come out of solution (degas) and present a possible explosion hazard. Examples of such equipment are:

- (a) the HT D<sub>2</sub>O storage tank
- (b) the bleed condenser

Both of these components have HT  $D_2O$  liquid in thermal equilibrium with  $D_2O$  vapour above the liquid.

In the D<sub>2</sub>O storage tank the cover gas is helium but  $\rm H_2/D_2$  gas will also be present due to degassing. A high D<sub>2</sub>/H<sub>2</sub> concentration ( $\sim 48$ ) means that purging is necessary to reduce the explosion hazard. Figure 2 shows typical H<sub>2</sub>/D<sub>2</sub> concentrations in the D<sub>2</sub>O storage tank corresponding to the H<sub>2</sub> gas addition flow rate changes. An increase of H<sub>2</sub> concentration in the storage tank as the HT H<sub>2</sub> addition rate is increased is evident. Purging the storage tank on high or rapidly increasing H<sub>2</sub>/D<sub>2</sub> concentrations is used to reduce the concentration as shown.

In the bleed condenser the 'cover' gas is mainly saturated D<sub>2</sub>O vapour with some D<sub>2</sub>/H<sub>2</sub> gases and also fission product gases such as Xe and Kr. As these (non condensable) gases build up in the D<sub>2</sub>O vapour space the heat transfer of the reflux cooling coils will be reduced and so the vapour temperature rises above that of liquid temperature. At a certain T the gases may then be degassed into the off gas system if one is installed. In practice if the off gas system is used to degas the bleed condenser it is done mainly to reduce the fission product gas activity rather than to release D<sub>2</sub>/H<sub>2</sub> gases which pose little explosion hazard in the wet atmosphere of the bleed condenser.

# 233.30-7

# (c) H2 Gas Addition System Shutdown

The  $\mathrm{H}_2$  addition system is not used when the reactor is shutdown and will be isolated whenever this is the case. In these circumstances there is reduced radiolysis, because the reactor is not operating and isolating will prevent an excess of  $\mathrm{H}_2$  building up in the HT system.

# (d) Changing of the H2 Cylinders

It should be noted that the  $\rm H_2$  gas cylinders will be declared spent, and changed, when the cylinder pressure drops to  $\sim 9$  MPa (HT system pressure). Hence the spent cylinders are not empty, but still contain  $\rm H_2$  gas at  $\sim 9$  MPa.

To avoid this problem of incomplete usage of the  $\rm H_2$  in the cylinders Bruce A and B are currently (1980) proposing to add the  $\rm H_2$  to the feed pump suction (purification outlet). The disadvantage of this would be the risk of pump gas locking, on too high an addition rate.

## ASSIGNMENT

- l. Compare your own plant's  ${\rm H}_2$  gas addition flowsheet with Figure 1 and explain any differences.
- 2. In your own plant how many  $H_2$  bottles/week are used in this system?
- 3. In your own plant, is the  ${\rm H_2}$  added by manual adjustment or by automatic control?
- 4. If there is a small or large LOCA in the HT system what should be done to the HT  $\rm H_2$  gas addition system? Explain.

D.J. Winfield